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Next-generation Surface Enhanced Raman Scattering (SERS) Substrates for Hazard Detection

by Mikella E. Farrell, Ellen L. Holthoff and Paul M. Pellegrino

ARL-RP-0411 **September 2012**

A reprint from *SPIE Defense and Security*, 23 April 2012

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We will report on the characterization and sensing capabilities of these next generation SERS substrates for the detection of energetic materials (ammonium nitrate, TNT, PETN, and RDX). Additionally, new efforts producing highly uniform samples, with known concentrations of energetic materials inkjet printed onto the SERS sensing surface using a precisely calibrated MicroJet system will be shown.

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KEYWORDS: SERS, sensor, chemical, biological, energetic, Raman, detection, ammonium nitrate

I. INTRODUCTION

As of early 2012, the world population was estimated to have increased to around 7 billion people. It is anticipated that between 50 and 70 million people will be added annually to the world population until the mid 2030s.[1] To feed this ever increasing population and demonstrate healthy economic growth, agricultural industries since the 1900's have relied on manufactured fertilizers to increase their agricultural output. One ubiquitous fertilizer component is ammonium nitrate (AN). Ammonium nitrate is a solid fertilizer (~ 34 % nitrogen) typically applied and distributed in a solid prilled form. Nitrogen-based fertilizers account for almost 97% of the fertilizers used in agriculture.[2] The forecast is for world nitrogen fertilizer demand to increase at an annual rate of about 1.4% until 2011/2012, which is an overall increase of 7.3 million tons.[1]

While AN has many practical applications in agriculture, it is also extensively used in the manufacture of industrial explosives. One common industrial use of AN is in the fabrication of ammonium nitrate fuel oil or ANFO-based explosives. A typical ANFO consists of 84% prilled ammonium nitrate that acts as an oxidizing agent and absorbant for fuel.[2] While there are many examples of AN being used in industry, unfortunately it is also becoming a regularly used and almost established component of fertilizer bombs or improvised explosive devices (IEDs) seen throughout the world. One tragic example of terrorist use of AN occurred in the attacks in the US in 1995 with the bombing of the Alfred P. Murrah Federal Building in downtown Oklahoma City.[3] Since then, the use of AN by terrorist groups and the resulting devastation from material detonation, has continued to challenge military and law enforcement personal both at home and abroad. In 2012, USA Today reported[4] that in Afghanistan the "number of improvised explosive devices that were cleared or detonated rose to 16,554 from 15,225, an increase of 9%." With the evident increase in IEDs and easily obtained and prevalent supply of ammonium nitrate in agricultural areas, the US Army and first responder community

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are increasingly focusing efforts on energetic materials detection and identification.[5-13] Analytes of interest include both energetic materials and commonly used components in IEDs.[14-16]

There are several examples of laser-based systems that were designed to allow rapid detection and identification of hazardous materials.[17] Many of these systems offer standard detection capabilities on new sensing platforms, like vehicle and man portable systems. Of the many available standoff techniques researched for energetic and hazardous materials detection, some that are gaining growing attention are Raman and Raman-based technologies.[18-23] Raman and Raman-based techniques are increasingly being used to address the outstanding need and challenge for rapid, sensitive, accurate detection, identification, and quantification of chemical, biological, and energetic hazards in many fields of interest (e.g., medical, environmental, industrial, and defense applications).[24] In particular, there has been s push to detect trace levels of hazardous materials left behind on common items like hair, clothes, and even fingerprints.

One spectroscopic detection method that is gaining significant popularity for meeting these sensitivity needs is surface enhanced Raman scattering (SERS). SERS-based techniques and platforms combine traditional spectroscopy with trace level detection capabilities. SERS is particularly advantageous and an appropriate detection technique to utilize for hazard detection as it does not suffer from interferences from water, requires little to no sample preparation, is robust and can be used in numerous environments, is relatively insensitive to the wavelength of excitation employed and produces a narrow-band spectral signature unique to the molecular vibrations of the analyte.

Since the discovery of SERS in the 1970's, it has been experimentally shown that the SERS signal enhancements are typically observed on metalized (typically silver or gold) nanoscale roughened surfaces.[25-27] In the SERS literature, the two mechanisms that have been shown to control the enhancement of the Raman scattering (SERS) are considered to be the electromagnetic fields generated at or near nanoparticle surfaces and the physical (chemical) adsorption of a target analyte onto a surface.[28-31] The electromagnetic enhancement (EM) is typically deemed to be the stronger contribution, with an enhancement factor (as compared to spontaneous Raman) ranging from approximately 10^4 to 10^{14},[32] while the chemical enhancement (CE) has been suggested to contribute at most 10^2.[33] Because of the many sensing advantages of SERS-based techniques, significant research efforts (defense, industrial and academic) have been directed toward developing SERS substrates for SERS-based sensor platforms.[33-42]

Due to the many advantages offered by SERS, there has been an intense research push to fabricate "better" SERS sensing platforms. Some of these SERS platforms are fabricated from colloids[43], film over nanospheres[44-47], fiber optic bundles[48], nanoparticles,[40, 49-52] and lithographically[53] produced structures. At best, typically the more sensitive substrate platforms generally have a 15% relative standard deviation (RSD; the measure of the reproducibility of an analysis) from substrate-to-substrate and SERS signal enhancements of 7 to 8 orders of magnitude.[45, 46] Consequently, many researchers in academia, industry, and government have focused concerted efforts toward increasing the signal enhancement ability, reproducibility, and mass production manufacturing of substrates to increase the utility of this technique.[53, 54] For the Army and first responders, such a substrate platform with increased sensitivity and reliability would be very advantageous for the detection and identification of unknowns.[55, 56]

Significant research efforts have also been concentrated on better directing the optimization of the substrate surface from which the SERS enhancement occurs.[53, 57-60] Based in part on experimental and theoretical efforts, the directed fabrication of SERS platforms has focused on modifying the feature size[61-63], spacing between objects, geometry and shape of structures[64-66], identity and incorporation of metals on the surface[67], feature height, and the character of the foundation layer [44, 68-70] on which the architecture is fabricated.[56, 71, 72] There are numerous examples in the literature detailing how variation in some of these parameters in certain cases can result in very dramatic changes to the overall SERS enhancing capabilities of the substrate surface. Rigorous efforts continue to focus on developing an understanding about how these parameters can synergistically work together to result in a highly reproducible and sensitive SERS substrate. As focus continues on improving the overall sensing capabilities of the SERS surface, congruently research continues to push towards the development of a uniform reproducible mass produced platform necessary to facilitate widespread incorporation of SERS in viable sensing platforms.

Some success fabricating both spectrally and physically reproducible SERS substrates has been demonstrated with commercially available standard Klarite™ substrates (Renishaw Diagnostics.).[58, 73-76] These substrates were developed using Silicon-based semiconductor fabrication techniques.[76] Klarite substrates are fabricated using a well defined

silicon fabrication technique in which a silicon diode mask is defined by optical lithography, and then potassium hydroxide (KOH) surface etched. The process results in an array of highly reproducible inverted pyramid features.[76] These array pyramids are reported to have "hot spots" or "trapped plasmons" located inside the wells.[76] These substrates have been previously characterized.[77] Atomic force microscopy (AFM) images have demonstrated that the inverted pyramid features are about 1.47 um in width and 1 um in depth. These substrates have plasmon absorbance bands located at 577 nm and 749 nm, thus demonstrating the usefulness of this substrate for a range of excitation sources. Additionally, due to the fabrication process used, under ideal conditions these substrates have demonstrated typical RSDs ranging from 10-15% under drop and dry conditions. While these standard Klarite substrates do demonstrate a high degree of substrate reproducibility and very low substrate background (SERS signal and surface morphology), for many applications to real-world situations, increased analyte sensitivity is still necessary. Recently new prototype Klarite based substrates have been fabricated by Renishaw Diagnostics with the intent to expand substrate sensing capabilities. The morphologies of these substrates dramatically differ in overall shape, pitch and spacing as compared to the standard Klarite substrate, resulting in very interesting sensing capabilities.[78, 79]

In this proceedings paper, preliminary efforts characterizing energetic materials deposited onto commercially available SERS substrates will be discussed and demonstrated. These hazard materials are deposited at low concentrations, serving as a proof of principle demonstration for future application for trace detection. By demonstrating these preliminary results, the potential of these SERS substrates for future incorporation in fielded Raman systems for the sensing of hazard materials can be imagined.

II. EXPERIMENTAL

Chemicals and Reagents. All chemicals were used as received without further purification. The common chemical ammonium nitrate (AN) was used as received from Sigma. The energetic samples 1,3,5-trinitro-1,3,5-triazine (RDX), trinitrotolulene (TNT) and pentaerythritol tetranitrate (PETN) were obtained from Cerilliant (10 mg/mL, 99.9% purity). All inkjet printer stock solutions were prepared in acetonitrile or a water and methanol containing solution.

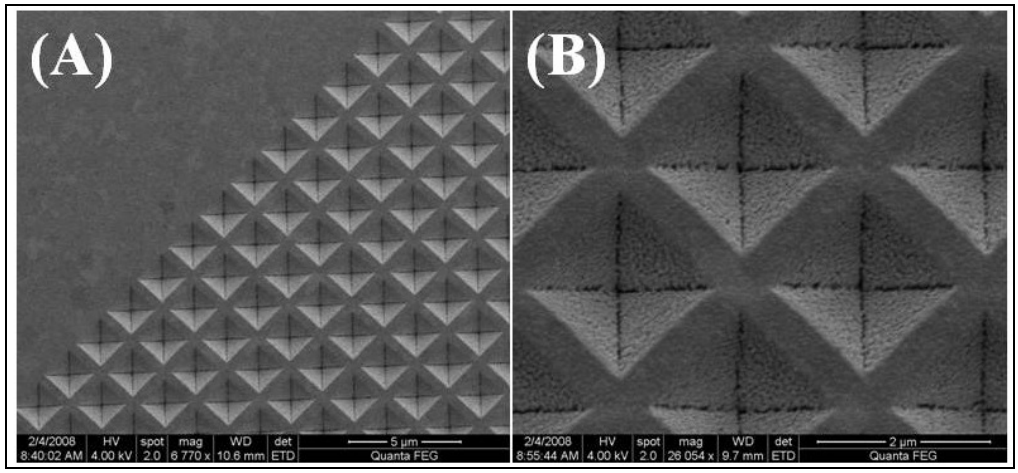

Figure 1. SEM data for the standard Klarite substrate as shown at two different magnifications. In (A) the active area is shown with the features and inactive area appears smooth. In (B) a higher magnification SEM image of the features of the substrate are shown.

Commercially available substrates. Commercially available slide mounted standard Klarite™ 302 SERS substrates were purchased from Renishaw Diagnostics. Slides were individually wrapped and vacuum sealed. The SERS active area on these slides is a small 4 mm x 4 mm wafer with a gold surface. The standard Klarite™ slides were only used once and opened just prior to measurement to reduce any possible surface fouling. Additionally, the substrate was submerged in ethanol to remove any possible contamination that may have accumulated on the surface. Next generation Klarite substrates (designated as 308's) were used as received from Renishaw Diagnostics following the same procedures used for the standard substrates. Most data in this proceedings paper will be presented as an average of a collected data set

and the standard deviation error shown, unless otherwise indicated. See Figure 1 for an example SEM image of a typical Klarite standard SERS substrate.

Instrumentation and Data Analysis.
Scanning electron microscope (SEM) images was obtained using a FEI environmental SEM (Quanta 200 FEG).

A Renishaw inVia Reflex Raman microscope was used for SERS and Raman spectra collection. Spectra were collected using a 785 nm laser. The laser light was focused onto the sample using a 5X objective, exposures were 10 seconds in length, and 3 accumulations were collected per spot. Approximately 7 mW of power irradiated the surface of the substrate. Five spectra were collected from each substrate. Samples were positioned using a motorized XYZ translational stage internal to the microscope. Spectra were collected, and the instrument was run using Wire 3.2 software operating on a dedicated computer. Data analysis was achieved using IgorPro 6.0 software (Wavemetrics).[78]

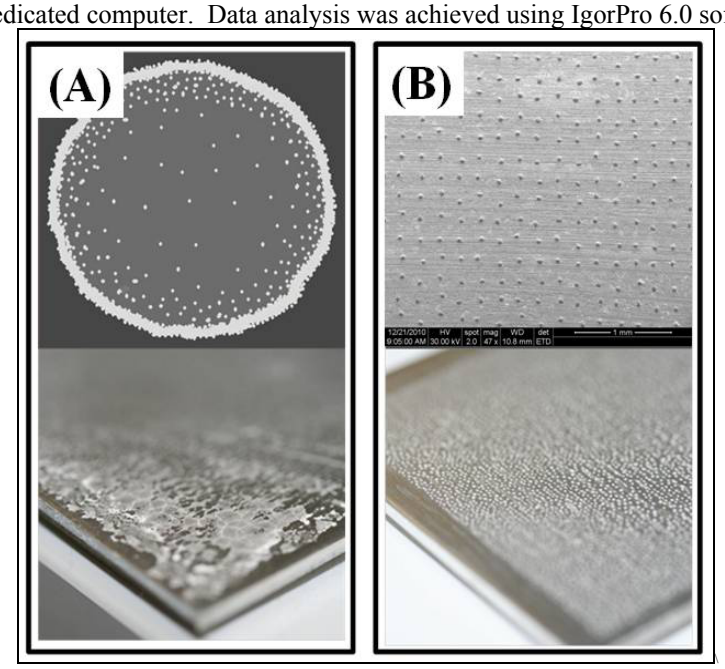

Figure 2. Example images from drop and dry and microjetted samples. (A) Graphic and photograph of drop and dry sample deposition demonstrates uneven dispersion and the coffee ring effect. (B) SEM image and photograph of an array of microdroplets demonstrates a more even sample dispersion when sample is dispensed with the microdispenser.

MicroFab Technologies Printer. Materials were produced using a JetLab® 4 (MicroFab Technologies) tabletop printing platform and have been previously documented.[80] Briefly, the JetLab® 4 is a drop-on-demand inkjet printing system with drop ejection drive electronics (JetDrive™ III), pressure control, a drop visualization system, and precision *X*, *Y*, *Z* motion control. The dispensing device (print head assembly, MJ-AL-01-060) consists of a glass capillary tube, with a 60 µm diameter orifice coupled to a piezoelectric element. Voltage pulses (20–25 V; rise time 1 µs; dwell time 28–32 µs; fall time 1 µs) applied to the piezo result in pressure fluctuations around the capillary. These pressure oscillations propagate through the printing fluid in the tube, resulting in ejection of a microdrop. Drops are visualized using synchronized strobe illumination and a charged coupled device (CCD) camera. Determining optimal jetting parameters is a trial-and-error process. Stable droplet ejection is achieved by visually observing expelled microdrops and adjusting voltage pulse parameters and capillary fluid backfill pressure. Conditions that provide the highest drop velocity without satellite droplet formation are desired. Printing was performed at a frequency of 250 Hz with a droplet velocity of ~2 m/s. Drop diameter was estimated to be ~60 µm, based on the capillary orifice diameter. During printing, a single substrate was placed on the sample stage. The print head remained fixed at a specified height while the stage moved to print a specified pattern. A rectangular array, which covers a rectangular area with rows of equidistant points, was pre-programmed based on the substrate size and desired sample concentration. An array pattern was chosen for the purpose of creating the effect of a homogeneous coating for optical interrogation. Depending on the desired concentration per unit area (e.g., $\mu g/cm^2$), the total number of drops needed to achieve the desired concentration in that area was calculated

based on the mass of a single microdrop. Based on the number of total drops needed, the array spacing and drops needed per line can be calculated. These values are easily adjusted depending on concentration. Arrays were printed using the print on-the-fly mode. In this mode, the stage moves continuously as a single microdrop is dispensed at each array element. Print on-the-fly mode improves sample throughput.

III. RESULTS AND DISCUSSION

Using the microjetting system, the analyte ammonium nitrate (AN) was jetted onto standard Klarite substrates at differing concentrations ranging from 0.05 ug/cm^2 to 50 ug/cm^2. These concentrations were independently validated by UV-Vis measurements (data not shown), and have been previously discussed.[80] This concentration range was initially chosen as a proof-of-principle demonstration to determine if the AN could be detected via SERS and validate that there was a change on overall SERS signal intensity with an increase in analyte concentration on the standard Klarite substrate. In Figure 3, some example SEM images (A-D) and a typical spectrum for AN are shown (E). In the SEM images in Figure 3 A and B, the images of the AN droplet physically appear dramatically different, (A) is smooth while (B) appears to have a faceted structure. It has previously been observed that at varying concentrations (trace vs. bulk) different phases of energetic materials can exist,[8, 21, 23, 81] however, this discussion is beyond the scope of this proceedings paper. In Figure 3C, AN printed at a 50 ug/cm^2 concentration is shown, and in 3D AN printed a 5 ug/cm^2 concentration is shown. In Figure 3E, a typical SERS spectrum of AN at a 50 ug/cm^2 concentration is shown, main bands are located at 713cm^{-1} and 1044 cm^{-1}. In the spectrum, the 1044 cm^{-1} band observed corresponds with the symmetric stretch mode of NO_3^-.[20, 82] For data analysis, the main AN band at 1044 cm^{-1} will be used for determining signal-to-noise (S/N) ratios.

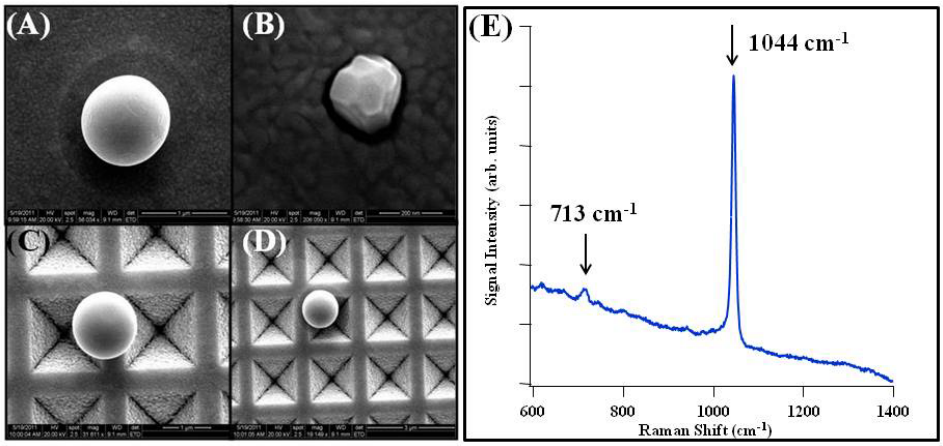

Figure 3. Example SEM images of ammonium nitrate deposited on a standard Klarite substrate. In (A) and (B) example physical difference in droplets observed are demonstrated. In (C) and (D) different concentrations of ammonium nitrate are deposited. In (E) a typical spectrum for AN is shown, main bands located at 713 cm^{-1} and 1044 cm^{-1}.

In Figure 4, data sets for AN deposited at different concentrations on Klarite substrates are shown. In Figure 4A, Raman and SERS data collected on a standard Klarite with AN concentrations of 5 ug/cm^2, 20 ug/cm^2 and 50 ug/cm^2 are shown. This data was analyzed comparing the S/N ratio for the 1044 cm^{-1} band of AN. From Figure 4A, it can be clearly seen that the SERS signal as compared to the Raman signal on a standard Klarite substrate is significantly greater. In Figure 4A, the 5 ug/cm^2 Raman S/N ratio is 2.02 and for SERS it is 38.75, 20 ug/cm^2 Raman S/N ratio is 5.48 and for SERS it is 67.50, and 50 ug/cm^2 Raman S/N ratio is 1.89 and for SERS it is 82.90. Thus it can be seen that the SERS signal is typically at least 19X greater than the Raman signal. Next, the detection capabilities of the standard Klarite and Klarite 308 were compared for trace levels of deposited AN. In Figure 4B, the SERS signal from AN deposited at 0.05 ug/cm^2 and 0.50 ug/cm^2 were compared on the two different SERS substrate types. For AN deposited at a concentration of 0.05 ug/cm2, the Raman S/N ratio is 1.24, the standard Klarite S/N ratio is 6.31, and the next generation Klarite 308 S/N ratio

is 17.28. This indicates that overall, the 308 is performing 2.73X better as compared to the standard commercial substrate. For AN deposited at a concentration of 0.50 ug/cm2, the Raman S/N ratio is 1.15, the standard Klarite S/N ratio is 9.03, and the next generation Klarite 308 S/N ratio is 32.17. This indicates that overall, the 308 is performing 3.6X better as compared to the standard commercial substrate. Thus indicating that for trace analysis, the next generation Klarite substrate performs better as compared to the standard Klarite substrate. With additional optimization, and the use of other energetic analytes, the overall signal increase with the next generation Klarite substrate may continue to improve.

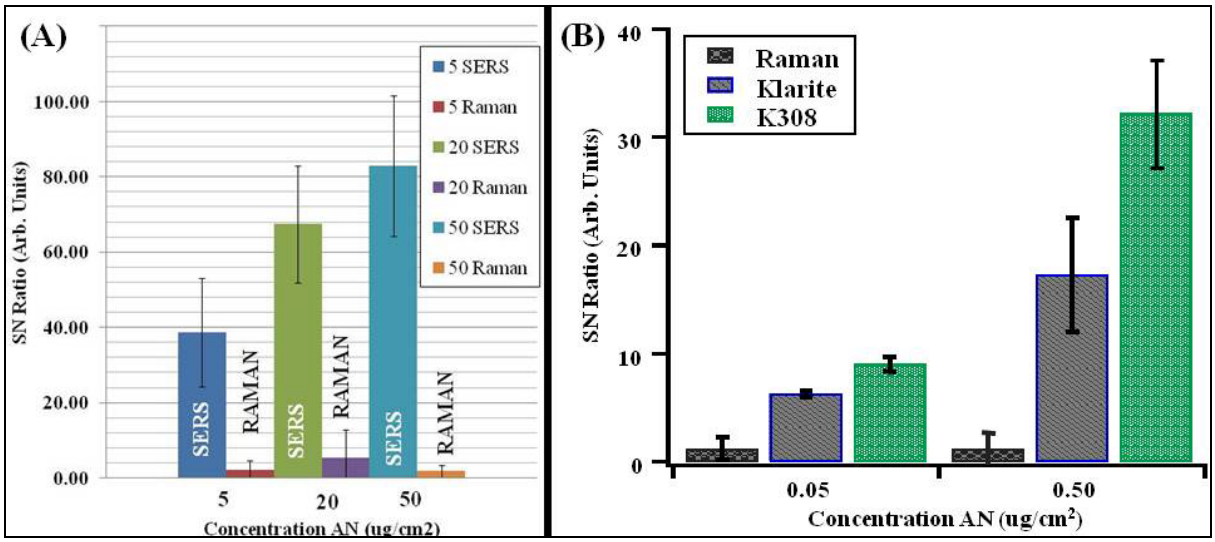

Figure 4. (A) S/N ratios for AN deposited at concentrations of 5, 20, and 50 ug/cm^2. SERS and Raman data is shown. In (B) the S/N ratios for standard and next generation Klarite substrates shown for AN deposited at 0.05 and 0.50 ug/cm^2.

Having successfully demonstrated the use of the standard and next generation substrates for sensing trace levels of IED components, we next demonstrated the applicability of using these substrates for detecting explosive samples. Explosives can be categorized as either primary or secondary based on how they are initiated. Primary explosives are typically detonated by some sort of ignition spark, and in some cases can be more challenging to handle. Secondary explosives require a detonator or primary explosive, and are typically characterized as being less sensitive to handling.[5] Some familiar examples of secondary explosives include RDX, PETN and TNT. Some of these secondary explosives can all be characterized as being very powerful and brisant.[83]

For these experiments, the energetic materials to be characterized with the commercial substrates include RDX, PETN, and TNT. These materials were microjetted onto standard Klarite substrates at a concentration of 5.0 ug/cm^2. The concentration jetted was validated using a secondary UV-Vis technique.[20, 81] After jetting, the samples were stored covered until just prior to use. In Figure 5A-D example SEM images of the explosive material RDX deposited on a standard Klarite substrate are shown. From Figure 5A, the low magnification SEM image demonstrates how the drop spreads out across the standard Klarite substrate, resulting in a minor "coffee ring" effect. In Figure 5D, a higher magnification SEM image clearly demonstrates the spread of energetic materials across the features of the standard Klarite substrate. In Figure 5D, example SERS spectra from a 5.0 ug/cm^2 concentration of TNT, PETN and RDX as collected on individual standard Klarite substrate are shown. The explosive material TNT has SERS bands located at 793 cm^{-1}, 823 cm^{-1} (NO_2 scissoring vibration) , and 1356 cm^{-1} (NO_2 symmetric stretching coupled to CN stretching).[20] PETN has main SERS bands observed at 869 cm^{-1} (ON stretch) and 1290 cm^{-1} (NO_2 symmetric stretch with some minor contributions from CH bending, CH_2 wagging and C_5 skeletal vibrations).[20] SERS bands for RDX are located at 877.4 cm^{-1} (symmetric ring-breathing mode) and 1315 cm^{-1} (NN stretch and CH_2 twist).[20, 84] From Figure 5D, it is clear that all explosives materials have very distinct and identifiable SERS bands that can be measured using the standard Klarite SERS substrates.

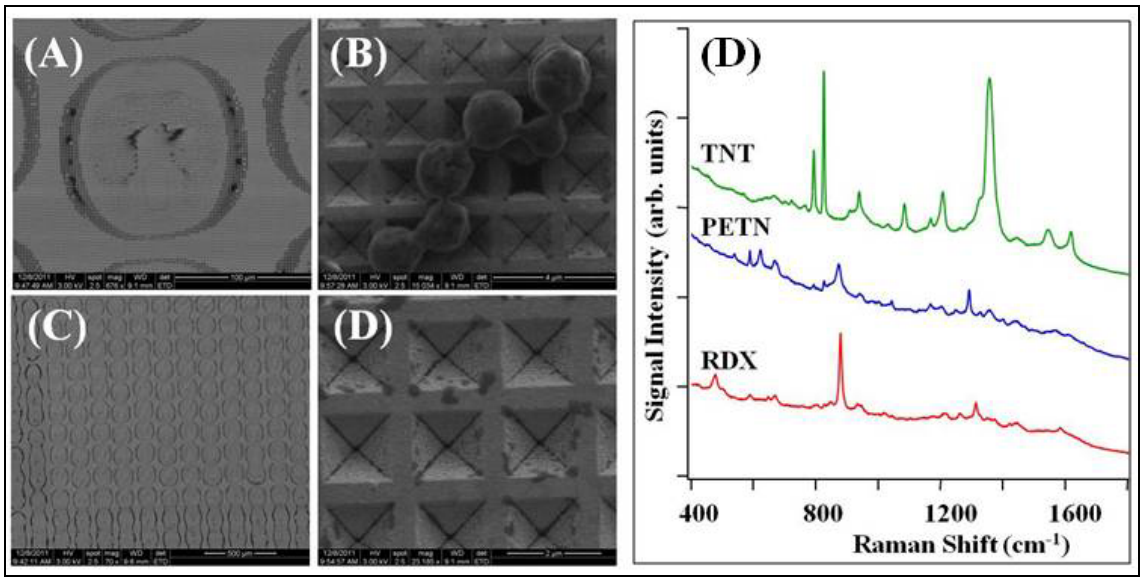

Figure 5. In (A-D) are example SEM images of RDX deposited on a standard Klarite substrate at different magnifications. In (E) example SERS spectra for common energetic samples TNT, PETN and RDX are shown.

Next, to demonstrate the potential for using the next generation Klarite substrates for SERS detection of low levels of energetic materials, the SERS spectra of the explosives TNT, PETN and RDX were collected. In these experiments the energetic materials were inkjetted onto different substrate surfaces at a concentration of 1 ug/cm^2. For data analysis via a S/N ratio determination, the 1356 cm^{-1} band of TNT, the 1290 cm^{-1} band of PETN, and the 877.4 cm^{-1} band of RDX were analyzed. In Figure 6A, example spectra for the energetic samples jetted onto the next generation Klarite substrate are shown. Bands used for S/N ratio determination are called out with a black arrow. In Figure 6B, the S/N ratios for the different energetic are shown (Raman and SERS measurements). From this data it was determined that at a 1 ug/cm^2 concentration the S/N ratio for TNT is 150.0±22.5 (Raman 4.2±1.4), for PETN is 41.7±8.4 (Raman 8.1±1.0), and for RDX is 40.3±7.1 (Raman 3.6±9.4). Analyzing this data it is determined that the next generation Klarite substrates can be used for the detection of low levels of energetic materials. Future work will determine the limits of detection possible with these substrates for energetic sensing.

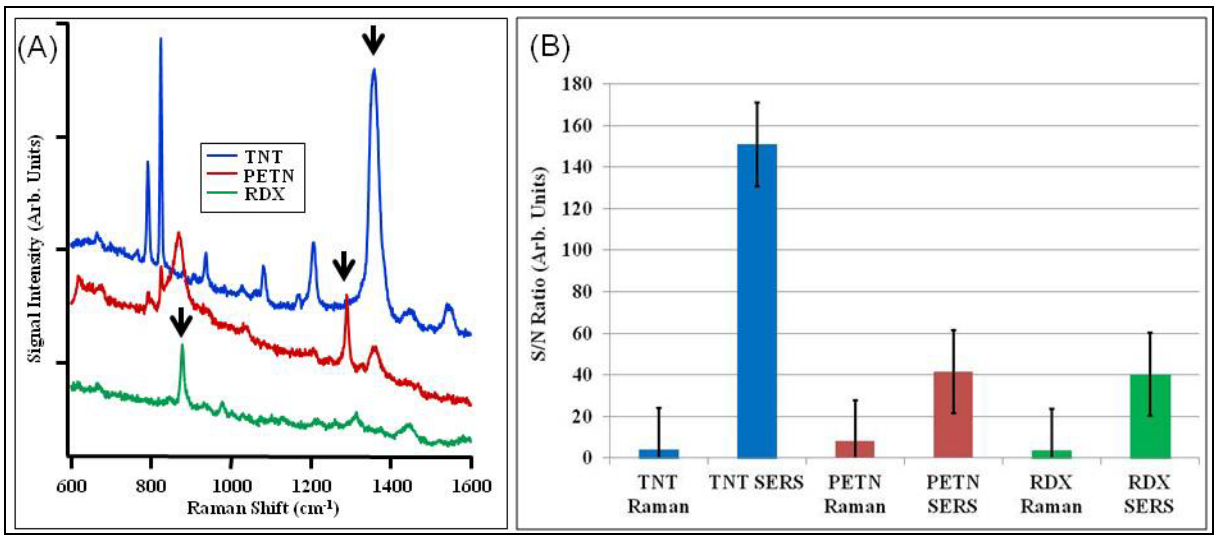

Figure 6. In (A) the spectra for TNT, PETN, and RDX as measured on the next generation Klarite 308 substrate. In (B) the S/N ratios for 1 ug/cm^2 of PETN and RDX inkjetted onto the active and nonactive areas of a next generation 308 Klarite substrate are shown.

IV. CONCLUSIONS

From the preliminary results shown, it can be concluded that the commercially available SERS substrates offer a viable approach for energetic detection at trace levels as demonstrated by detection of the common energetic hazards TNT, PETN, and RDX. Additionally, viability of detection of the common IED component AN was demonstrated. Future studies will include characterizing the response of the next generation Klarite substrates for trace samples of hazard materials.

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